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DOI:
10.1080/00268970903535483

Citation for published version (Harvard):

Link to publication on Research at Birmingham portal

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Download date: 23. Oct. 2023
Vacuum-UV negative photoion spectroscopy of CH₄


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Using synchrotron radiation in the range 12–35 eV, negative ions are detected by mass spectrometry following vacuum-UV photoexcitation of methane. Ion yields for H\(^-\), CH\(^-\) and CH\(_2\)\(^-\) are recorded, the spectra of CH\(^-\) and CH\(_2\)\(^-\) for the first time. All ions display a linear dependence of signal with pressure, showing that they arise from unimolecular ion-pair dissociation. Cross sections for ion-pair formation are put onto an absolute scale by calibrating the signal strengths with those of F\(^-\) from SF\(_6\) and CF\(_4\). Following normalisation to total vacuum-UV absorption cross sections, quantum yields for anion production are reported. There is a major discrepancy in the H\(^-\) cross section with an earlier measurement, which remains unresolved. The anions arise from both direct and indirect ion-pair mechanisms. For a generic polyatomic molecule AB, the former is defined as AB \(\rightarrow\) A\(^-\) + B\(^+\) (+ neutrals), the latter as the predissociative crossing of an initially-excited Rydberg state of AB by an ion-pair state. In a separate experiment, the threshold photoelectron spectrum of the second valence band of CH\(_4\), ionisation to CH\(_4\)^\(^+\) \(\tilde{A}^2A_1\) at 22.4 eV, is recorded with an instrumental resolution of 0.004 eV; many of the Rydberg states observed in indirect ion-pair formation converge to this state. The widths of the peaks are lifetime limited, increasing with increasing \(v\) in the \(v_1\) (a\(_1\)) vibrational ladder. They are the first direct measurement of an upper value to the dissociation rate of these levels into fragment ions.

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Keywords: methane; ion-pair formation; vacuum-UV; absolute cross sections; quantum yields.
1. Introduction
The interaction of electromagnetic radiation with the prototypical molecule, methane, is of fundamental interest because it is central to organic chemistry and abundant in our upper atmosphere. Much previous photochemical analysis has therefore been performed on the molecule. Being closed-shell and very stable, the lowest-lying excited electronic states of methane lie at high energies above the ground state, in the vacuum-UV region. As a consequence, much of the atmospheric photochemistry of methane is driven by intense solar atomic emission lines, such as Lyman-α radiation [1,2].

Methane is a tetrahedral molecule with the electron configuration: \((1a_1)^2(2a_1)^2(1t_2)^6\), and photoionisation has been studied by He I and He II Photoelectron spectroscopy [3-9]. The removal of the \(1t_2\) electron, at \(ca.\ 13\) eV, is known to give rise to the triply-degenerate ground state of \(\text{CH}_4^+\), and much detailed work has been carried out to study the nature of the Jahn-Teller distortion of the cation from tetrahedral symmetry [4,10,11]. The \((2a_1)^{-1}\) band at \(ca.\ 22\) eV gives rise to a long vibrational progression in the \(v_1 (a_1)\) mode [8], and the \((1a_1)^{-1}\) core excitation at \(ca.\ 290\) eV has been studied with vibrational resolution, revealing a shorter \(v_1\) vibrational progression, reviewed in [12]. Recent experimental studies have been undertaken to investigate the non-Franck-Condon behaviour of this core photoexcitation [13,14]. Between the \((2a_1)^{-1}\) and the \((1a_1)^{-1}\) ionisation energies (IEs), weak satellite peaks have been observed in the photoelectron spectrum, which have been assigned as ‘double-hole one-electron’ states of \(\text{CH}_4^+\) [8]. The formation of doubly-excited states of methane by photoexcitation, which correspond to Rydberg states that converge on these satellite states of \(\text{CH}_4^+\), have been investigated by dispersed fluorescence [15,16].

There have been many studies of the dissociation products following ionisation of \(\text{CH}_4\), revealing the fragmentation dynamics of an energy-selected \(\text{CH}_4^+\) cation [7,9,17-19], and detailed Lyman-α photofragmentation studies have also been undertaken [2,20]. Total photoabsorption cross sections have been measured in the vacuum-UV range of 10-30 eV many times [21-25], with cross sections ranging from \(ca.\ (1 – 5) \times 10^{-17}\) cm\(^2\).
In this paper we describe an experiment to detect anions following vacuum-UV excitation as a means to study the decay dynamics of electronically excited states of CH\textsubscript{4} due to ion-pair formation, generically described as AB → A\textsuperscript{−} + B\textsuperscript{+} (+ neutrals). Absolute cross sections and quantum yields have been evaluated for the anions observed. Photoion-pair formation has been detected for many diatomic and small polyatomic molecules [26]. Dissociative ion-pair states can be accessed via direct ion-pair formation, i.e. photoabsorption from the ground state directly to the ion-pair state, or via pre-dissociation following photoexcitation to an excited neutral state, often a Rydberg state. For the latter process to occur, the excited state must be formed at an energy greater than or equal to the asymptotic energy for ion-pair formation, and significant coupling between the two unperturbed wavefunctions is required [27].

The formation of H\textsuperscript{−} from CH\textsubscript{4} has previously been investigated in the energy range 12–27 eV by Mitsuke et al. [28,29]. In this paper, we confirm the H\textsuperscript{−} efficiency curve features that they detected, and extend the range of detection to higher energy. However, there is a significant difference in our value for the absolute cross section for H\textsuperscript{−} formation from that quoted by Mitsuke et al. We also report the first observation of the CH\textsuperscript{−} and the CH\textsubscript{2}\textsuperscript{−} anions from vacuum-UV photoexcitation of methane, and present absolute cross sections for formation of these anions.

2. Experimental

The ion pair apparatus has been described in detail previously [30]. Briefly, a direct jet of the gas under investigation is injected from a needle which orthogonally bisects the incident photon beam. The crossing point is positioned between two grids along the third Cartesian axis. A potential difference applied across these grids attracts negative ions towards a three-element electrostatic lens for focussing, and into a Hiden Analytical HAL IV triple quadrupole mass spectrometer (QMS) for mass selection. Anion detection is achieved by a channeltron electron multiplier. The apparatus and QMS were connected via a 1 mm diameter aperture, and were pumped by separate turbomolecular pumps, backed by a common rotary pump. Differential pumping enhances sensitivity by reducing the number of free electrons and secondary collisions in the QMS. All measurements were performed using vacuum-UV radiation from beamline 3.1 at the UK Daresbury Synchrotron Radiation Source (SRS), using
a 1 m focal length Wadsworth monochromator to provide tunable radiation in the range 12–35 eV. These energies are available from the higher energy of two gratings mounted back-to-back in the monochromator [31]. The optimum resolution that can be obtained from this beamline is 0.05 nm, corresponding to ca. 0.01 eV at 15 eV. However, to enhance sensitivity, the spectra reported in this paper were recorded at a resolution of 0.25–0.60 nm. A 2 mm diameter, 300 mm long capillary light guide was used to connect the beamline to the experimental apparatus, providing the necessary differential pumping.

The base pressure of the apparatus was ca. 10^{-7} mbar. The pressure was measured in the main chamber using an ionisation gauge, and the introduction of the sample gas to the system raised the pressure to ca. 10^{-5} mbar. The sensitivity of the ionisation gauge to CH_4, SF_6 and CF_4, essential for determination of absolute cross sections of anion formation, was calibrated in a separate experiment relative to N_2 using a capacitance manometer [32]. Gas samples were obtained from Apollo Scientific with quoted purity of > 99.9 %, and were used without further purification.

Mass spectra were recorded to observe all anions produced from photoabsorption of the sample gas by exposure to white light, i.e. the grating is set to zero order to act as a mirror. The mass-to-charge ratio (m/z) of each peak in the mass spectrum was then defined, and the ion yield recorded as a function of photon energy. Once the peak positions were determined, the anion signal was recorded as a function of gas pressure over a typical range of ca. (0.5 – 5.0) x 10^{-5} mbar. Anions displaying a linear dependence with pressure can be attributed to ion-pair formation (i.e. AB → A^- + B^+ (+ neutrals)), whereas those showing a non-linear pressure dependence cannot. The latter are likely to result from the two-step kinetic process of dissociative electron attachment (i.e. AB + hν → AB^+ + e^-; AB + e^- → A^- + B), in which the rate of formation of A^- is proportional to the square of the pressure of AB.

The yields of H^-, CH^- and CH_2^- all show a linear dependence with pressure. To determine their absolute cross sections, it is necessary to normalise the signals to the photon flux, the ring current, the gas pressure, the ionisation gauge sensitivity, and the relative mass.
sensitivity of the QMS to detection of the different anions. As in our previous studies on 
SF₅CF₃ [33] and the CF₃X series [34], we can write that:

\[ \sigma(h\nu) = k \left( \frac{SM}{frp} \right) \]  \hspace{1cm} (1)

where \( S \) is the detected signal normalised to unit time, \( f \) is the relative photon flux which 
effectively is a measure of the grating efficiency, \( r \) is the storage ring current, \( p \) is the sample 
gas pressure corrected for ionisation gauge sensitivity, and \( M \) is the relative mass sensitivity 
of the QMS. \( k \) is the normalisation constant. Normalisation to \( f \), \( r \) and \( p \) is facile, but this is 
not the case for \( M \). An extensive set of experiments was therefore performed to determine \( M \) 
as a function of \((m/z)\), and is described in Section 2.1.

The normalised signals can then be put on an absolute scale by calibration with the F⁻ signals 
measured from SF₆ \((7 \pm 2) \times 10^{-21} \text{ cm}^2 \text{ at } 14.3 \text{ eV [35]} \) and from CF₄ \((1.25 \pm 0.25) \times 10^{-21}
\text{ cm}^2 \text{ at } 13.9 \text{ eV [36]} \). We note, however, that these cross section values from Mitsuke et al. 
are not strictly absolute, but are obtained indirectly from the signal of O⁻ produced from O₂ at 
17.3 eV, for which the absolute cross section is known [37]. The values of the normalisation 
constants, \( k (\text{F}^-/\text{SF}_6) \) and \( k (\text{F}^-/\text{CF}_4) \), should be equivalent, but in fact they differ by a factor 
of 1.5. Given the number of corrections made to the anion signals in the two experiments, this 
discrepancy falls within a reasonable expected experimental uncertainty. The average value 
of \( k \) was then used in Equation (1) to determine the absolute cross sections, \( \sigma \), with units of 
\text{cm}^2, for production of H⁻, CH⁻ and CH₂⁻ from CH₄.

2.1 Mass Discrimination by the QMS

All quadrupole mass spectrometers exhibit an element of mass discrimination due to fringing 
fields, with a tendency to transmit heavier ions less efficiently [38]. To correct for this effect, 
the mass factor, \( M \), has been determined by comparing the cation mass spectra of many 
polyatomic molecules in the QMS, following 70 eV electron impact ionisation, to actual mass 
spectra published in the electronic NIST database [39]. It is assumed that any mass factors in
the data presented in reference [39] have been accommodated. \( M \) was calculated as a function of \((m/z)\), see Figure 1, and was used to normalise the raw anions signals, as explained above. It can be seen that as \((m/z)\) increases, the detection efficiency of the QMS decreases and a higher \( M \) value is required to correct this effect.

The zero-blast artefact [38], whereby all ions entering the quadrupole mass filter may be transmitted when the applied potentials are set to detect \(m/z\) 1, is not important in this CH4 study because the \(H^-\) signal is dominant. This effect, however, is important in the detection of the weak \(H^-\) signal from \(CH_3X (X = F, Cl, Br)\), where the spectra show significant contributions from the much stronger \(X^-\) ion yields [40].

3. Thermochemistry

Our work determines appearance energies at 298 K (AE\(_{298}\)) for fragment anions from CH4, and these can be compared with calculated thermochemical values. Berkowitz has noted that for many polyatomic molecules, a calculated threshold energy is a lower limit to the experimental AE\(_{298}\) value of an anion, when suitable assumptions are made about what the accompanying cation and neutral fragment(s) are [26]. Furthermore, in comparing experimental AE values with calculated \(\Delta_r H^o\)\(_{298}\) values of appropriate dissociation reactions, we are making two assumptions. Firstly, the dissociated fragments are not initially formed with thermal equilibrium, but rather are produced with conserved translational momentum relative to the centre of mass. Therefore a well-defined thermodynamic temperature cannot be allocated to the moieties and thermal corrections should be made [41]. Secondly, the effects of entropy have been disregarded, whereas for unimolecular dissociative reactions \(\Delta_r G^o\)\(_{298}\) is always slightly more negative than \(\Delta_r H^o\)\(_{298}\) because \(\Delta n\) (the stoichiometric difference in the number of gas-phase molecules due to the reaction) and hence \(\Delta_r S^o\)\(_{298}\) are both positive. Both of these effects are ignored in this study, which is deemed justifiable at the relatively modest resolution of the experiment.

Values for \(\Delta_r H^o\)\(_{298}\) of relevant ion-pair reactions were calculated using literature values for enthalpies of formation (\(\Delta_r H^o\)\(_{298}\) in kJ mol\(^{-1}\)): CH4 = –74.9, H = 218.0 (both Ref. 42); H\(^-\) =
145, CH\textsuperscript{−} = 477, CH\textsubscript{2}\textsuperscript{−} = 327, H\textsuperscript{+} = 1530, H\textsubscript{2}\textsuperscript{+} = 1488, CH\textsuperscript{+} = 1619, CH\textsubscript{2}\textsuperscript{+} = 1386 and CH\textsubscript{3}\textsuperscript{+} = 1098 (all Ref. 43).

4. Results and Discussion

4.1 Ion-Pair Spectra

The ion yields of H\textsuperscript{−}, CH\textsuperscript{−} and CH\textsubscript{2}\textsuperscript{−} from CH\textsubscript{4} in the range 12–35 eV are shown in Figures 2(a) – 2(c), respectively. The resolution is 0.6 nm, corresponding to 0.07 eV at 12 eV and 0.28 eV at 24 eV. The H\textsuperscript{−} signal is the most intense. Since all three anions show a linear dependence of signal with pressure, it is possible to determine the absolute cross sections for anion production. Using absolute, total vacuum-UV absorption cross section data from Au et al. [24], quantum yields for anion production can also be determined. Table 1 shows these data for the energies at which the H\textsuperscript{−}, CH\textsuperscript{−} and CH\textsubscript{2}\textsuperscript{−} anions have maximum intensity; 20.6, 29.3 and 24.9 eV, respectively. A very weak signal was detected at m/z 15 (CH\textsubscript{3}\textsuperscript{−}), but the signal-to-noise ratio was poor, and the features appeared to mimic those in the CH\textsubscript{2}\textsuperscript{−} spectrum. It is difficult to differentiate weak signals between anions in the QMS that are only one m/z unit apart, and this spectrum was therefore discarded. Figure 2(d) shows the vibrationally-resolved threshold photoelectron spectrum (TPES) of the second band of CH\textsubscript{4}, i.e. ionisation to CH\textsubscript{4}\textsuperscript{+} ̃\textit{A}\textsuperscript{2}\textit{A}_1, recorded with the imaging photoelectron photoion coincidence (iPEPICO) spectrometer at the Swiss Light Source, Paul Scherrer Institute Villigen, Switzerland [44]. There is no similarity between this spectrum and any of the three anion yields over this energy range. In addition to the pressure test, this is further evidence that the anions are not formed by dissociative electron attachment, but by ion-pair dissociation.

4.1.1 Appearance energies and thermochemical thresholds

The arrows in Figures 2(a)–2(c) show the calculated $\Delta_{\text{f}}H^\circ_{298}$ values for ion-pair dissociation reactions (2) – (10). (As described earlier, we are not distinguishing a reaction enthalpy from a reaction energy). They take the following values; 13.66, 18.91, 19.06, 23.58, 21.58, 23.40, 26.10, 19.59 and 22.28 eV.

\[
\text{CH}_4 \rightarrow \text{H}^- + \text{CH}_3^+ \quad (2)
\]
The thermodynamic thresholds can be used to infer the possible decay channels that give rise to each peak in the spectrum. The AE$_{298}$ of H$^-$ precedes the calculated $\Delta_rH^o_{298}$ value of reaction (2) by ca. 0.4 eV. This scientific impossibility could be accounted for by uncertainties in the thermochemistry, and/or by contributions from hotbands. However, it is clear that the first peak in the H$^-$ spectrum at ca. 15 eV can only arise from reaction (2). The double peak in the H$^-$ spectrum at ca. 21 and 22 eV probably corresponds to either or both decay channels shown in reactions (3) and (4). The third broad peak at ca. 28 eV has an approximate onset at ca. 23 eV, and may correspond to production of H$^-$ by reaction (5).

\[
\begin{align*}
\text{CH}_4 & \rightarrow \text{H}^- + \text{CH}_2^+ + \text{H} \quad (3) \\
\text{CH}_4 & \rightarrow \text{H}^- + \text{CH}^+ + \text{H}_2 \quad (4) \\
\text{CH}_4 & \rightarrow \text{H}^- + \text{CH}^+ + 2\text{H} \quad (5) \\
\text{CH}_4 & \rightarrow \text{CH}^- + \text{H}^+ + \text{H}_2 \quad (6) \\
\text{CH}_4 & \rightarrow \text{CH}^- + \text{H}_2^+ + \text{H} \quad (7) \\
\text{CH}_4 & \rightarrow \text{CH}^- + \text{H}^+ + 2\text{H} \quad (8) \\
\text{CH}_4 & \rightarrow \text{CH}_2^- + \text{H}_2^+ \quad (9) \\
\text{CH}_4 & \rightarrow \text{CH}_2^- + \text{H}^+ + \text{H} \quad (10) \\
\end{align*}
\]

The CH$^-$ spectrum shows an onset at 22.5 ± 0.2 eV. A weak shoulder is observed up to ca. 27 eV, and a second, more prominent onset is apparent at this energy. The first onset could correspond to reaction (6) or (7), although we believe that the shoulder is probably an artefact of the spectrum, resulting from detection of the CH$_2^-$ anion of comparable intensity which is only 1 m/z unit apart from CH$^-$. Thus we propose that the true onset of CH$^-$ formation from CH$_4$ is ca. 27 eV. It is difficult to assign this second/true onset to a particular dissociation reaction, as it could correspond to reactions (6), (7) and/or (8). The displacements of the $\Delta_rH^o_{298}$ values of reactions (6) and (7) from the onset of this peak would be reasonable, due to the formation of a new H$_2$ or H$_2^+$ bond, respectively. An energy barrier resulting from
forming these bonds is extremely likely, and thus their thermochemical thresholds would be expected to lie below the experimental onsets. The onset at $22.2 \pm 0.2$ eV in the CH$_2^-$ yield agrees very closely with the enthalpy of reaction (10), but similarly the peak could correspond to reaction (9), with formation of H$_2^+$. 

4.1.2 The H$^-$ yield from CH$_4$

We now consider the H$^-$ spectrum (Figure 2(a)) in more detail. The line positions and relative peak intensities agree well with those observed by Mitsuke et al. [28,29]. The first peak at ca. 15 eV, with onset at $13.3 \pm 0.1$ eV, has been assigned as a direct transition to an ion-pair state which dissociates to into H$^-$ (1S) + CH$_3^+$ ($\tilde{X}^1A_1^+$) [28]. We note that this onset lies well above the first adiabatic IE of methane, 12.61 eV [4,5], and well below the second adiabatic IE, 22.39 eV [8]. The H$^-$ peak at 15 eV cannot, therefore, coincide with a Rydberg state of CH$_4$, and its broad shape and slow onset indicate direct ion-pair formation [45].

The region between 19.5–23.5 eV in Figure 2(a) contains fine structure that is shown in more detail in a higher-resolution spectrum (Figure 3). Peak positions are listed in Table 2. These features arise from an indirect process, in which an initially-excited Rydberg state predissociates into an ion-pair state. Thus the vacuum-UV radiation is probing the spectroscopic features of the Rydberg states. Vibrational structure is observed in three close-lying Rydberg states, and they have been assigned by Mitsuke et al. to the $(2a_1)^{-1}(3p)^1$, $(2a_1)^{-1}(4p)^1$ and $(2a_1)^{-1}(5p)^1$ states [28,29]. These Rydberg states converge on the $\tilde{A}^2A_1^+$ state of CH$_4^+$, so the vibrational progressions observed should mimic closely any vibrational structure in the second photoelectron band of CH$_4$. When an np Rydberg state is excited, the only allowed fundamental frequency in $T_d$ symmetry is the totally-symmetric $\nu_1$ (a$_1^+$) C–H stretching mode. We observe a vibrational spacing in the $(2a_1)^{-1}(4p)^1$ Rydberg state, in which the vibrational structure is most clearly resolved, of $0.26 \pm 0.02$ eV or $2097 \pm 160$ cm$^{-1}$, to be compared with a value for neutral CH$_4$ of 2917 cm$^{-1}$ [42]. As expected, this matches the $\nu_1$ vibrational spacing of CH$_4^+$ $\tilde{A}^2A_1^+$ of $0.27 \pm 0.01$ eV, taken from the threshold photoelectron spectrum of this band (Figure 4, data listed in Table 3).
A Rydberg series of energy levels, \( E_n \), is described by the well-known formula:

\[
E_n = IE - \left( \frac{R_H}{(n - \delta)^2} \right)
\]  

(11)

where \( R_H \) is the Rydberg constant, \( n \) is the principal quantum number of the orbital which runs to infinity upon convergence, and \( \delta \) is its quantum defect. \( \delta \) values for the 3p and 4p states are determined to be \( 0.62 \pm 0.02 \) and \( 0.58 \pm 0.02 \), to be compared with values of \( 0.67 \) and \( 0.60 \) from Mitsuke et al. [28]. There is some inconsistency in the literature whether to use the adiabatic or the vertical IE in such Rydberg calculations, and this choice can significantly affect the Rydberg assignments for large values of \( n \), near the convergence limit. As the Rydberg formula determines the electronic series of states, \( E_n \) (Equation (11)) should refer to the \( v''=0 \rightarrow v'=0 \) transition from the ground state to the Rydberg state, and the adiabatic IE should be used in Equation (11). However, it is only possible to do this if vibrational structure is resolved in the spectrum and the transition to \( v'=0 \) is identifiable. In spectra that consist of many unresolved vibrational modes, it is more appropriate to use the vertical IE, because both the vertical Rydberg and vertical ionisation transitions will occur from \( v''=0 \) to the same value of \( v' \). This makes the valid assumption that the geometry and vibrational spacing of the Rydberg and cation states are similar. In determining the quantum defects of these Rydberg states of CH₄, we have calculated all the \( v_R \rightarrow v''=v_R \) transitions per Rydberg state, using the vibrationally-resolved term values in Tables 2 and 3. The values for the 3p and 4p Rydberg states given above, \( 0.62 \pm 0.02 \) and \( 0.58 \pm 0.02 \), are average values. The quantum defect of the \( (2a_1)^{-1}(5p)^1 \) Rydberg state has not been determined because the \( v''=0 \rightarrow v'=0 \) transition is not categorically defined. We note that the same vibrational intensity distribution is not expected for the \( H^- \) yield produced by indirect ion-pair formation through the np Rydberg series as a photoelectron spectrum of the \( CH_4^+ \stackrel{\Delta}{\rightarrow} 2^2A_1 \) state; the former is an indirect two-step process, the latter is a direct one-step process. Indeed, a different intensity distribution is observed in the Rydberg 3p and 4p series (Figures 3 and 4).
Between 23.5 and 33.0 eV, Figure 2(a) displays an extensive peak which has not previously been observed. This peak lies above the second adiabatic IE (22.39 eV [15]), but well below the next IE, ca. 290 eV, which corresponds to the (1a1)−1 core ionisation. Thus, it features in an energy range absent of ‘single-hole-one-electron’ Rydberg states. In addition, the broad and structureless nature of the peak does provide some tangible evidence for a direct ion-pair process, and it could also result from a shape resonance. However, it seems more likely that this peak results from indirect ion-pair formation, following predissociation of many close, unresolved ‘doubly-excited Rydberg states’ that converge to a doubly-excited IE of methane. Doubly-excited states of methane have been observed by dispersed fluorescence in the 25–35 eV energy range by Kato et al. [15]. The states converge to the ‘double-hole one-electron’ states of CH₄⁺, observed as satellites in the photoelectron spectrum by Carlsson Göthe et al. [8]. Doubly-excited states were observed as a broad peak at ca. 29 eV in the study of Kato et al., which gave rise to fluorescence from atomic hydrogen. They were assigned as Rydberg states converging on the (1t₂)²(3a₁)¹ state of CH₄⁺ at 32.1 eV, that produce excited H atoms via predissociation [8]. Furthermore, it is possible that the broad peaks in the CH⁻ and the CH₂⁻ spectra (Figures 2(b) and 2(c)) at ca. 29 and 25 eV, respectively, also arise from indirect ion-pair formation, resulting from different ion-pair states crossing these same doubly-excited Rydberg states. We note that the signals of H⁻, CH⁻ and CH₂⁻ have all virtually disappeared at 32.1 eV. In addition, reactions (5), (8) and (10), which we believe to be the most likely routes for formation of these three anions at energies above 25 eV, all involve the production of neutral H atoms.

4.1.3 Absolute cross sections and quantum yields for anion formation

The absolute cross section for H⁻ formation, 1.4 x 10⁻²² cm² at the peak of the (2a₁)⁻¹(3p)¹ Rydberg state at 20.6 eV, is a factor of ca. 70 smaller than the value quoted by Mitsuke et al. at the slightly higher energy of 21.5 eV, the peak of the 4p Rydberg state [29]. Whilst the errors made in evaluating absolute cross sections are often underestimated in the literature, we do not believe that this major discrepancy can be explained by an accumulation of individual errors in the various correction factors described in Section 2. Overall, we believe that our cross sections are accurate to within a factor of 2–3. The corrections made by Mitsuke et al. to their signals to determine absolute σ values are not clear, and in particular it is not apparent whether they have applied any mass discrimination correction factor for
detection of m/z 1 anions in their quadrupole mass spectrometer. We therefore measured the 
H⁻ yield from C₂H₆ and C₃H₈, obtained absolute cross sections in the manner described 
earlier, and compared our data with that of Mitsuke et al. for these larger saturated 
hydrocarbons [29]. Rather surprisingly, the results are in reasonable agreement [46]. For 
example, at the energy at which the cross section is a maximum, 18.9 eV, we determine σ 
(H⁻/C₂H₆) to be 1.7 x 10⁻²¹ cm², to be compared with 2.2 ± 0.9 x 10⁻²¹ cm² from Mitsuke et 
al. For H⁻/C₃H₈, at the peak energy of 18.7 eV, we determine a cross section of 3.4 x 10⁻²¹ 
cm², to be compared with the value from Mitsuke et al. of 1.6 ± 1.0 x 10⁻²¹ cm². It appears, 
therefore, that the detection of m/z 1 anions is not the reason, per se, for the anomalously high 
value for σ (H⁻/CH₄) of Mitsuke et al.

We note that in comparing cross sections for H⁻ formation from CH₄ to F⁻ formation from 
CF₄, we might expect the H⁻ cross sections to be smaller on electronegativity grounds 
because the C–H bonds are polarised Cδ⁻–Hδ⁺, whereas the C–F bonds are polarised Cδ⁺–Fδ⁻ 
[3]. Assuming that the cross section for F⁻/CF₄ at 13.9 eV, 1.25 x 10⁻²¹ cm² [36], is correct, 
and indeed our data are calibrated to this value (Section 2), then it is surprising that the value 
of the cross section for H⁻/CH₄ from Mitsuke et al. is eight times greater than that for F⁻/CF₄, 
whereas our value is nine times smaller. We have also observed this trend in cross sections 
(i.e. σ (F⁻) > σ (H⁻)) for larger hydrocarbons and their perfluorinated equivalents; C₂H₆ 
(C₂F₆), C₃H₈ (C₃F₈) and C₂H₄ (C₂F₄). In each case, in the range 10–25 eV the maximum 
value of the F⁻ cross section is a factor of 2–18 times greater than the maximum value for H⁻ 
formation [46]. These arguments provide some evidence, and give confidence to our smaller 
value of σ (H⁻/CH₄) shown in Figure 2(a).

Using our cross section values and total absorption cross sections from (e,2e) spectroscopy 
[24], the absolute quantum yields for H⁻, CH⁻ and CH₂⁻ formation can be calculated. They 
take values in the range 1–5 x 10⁻⁶ (Table 1). These values are of the same order of magnitude 
as those obtained in our earlier studies on CF₃X (X = Cl,Br,I) [34] and SF₅CF₃ [33].

4.2 Threshold Photoelectron Spectrum of the second band of CH₄
The TPES of the second band of CH$_4$ ionisation to CH$_4^+$ $\tilde{A}$ $^2A_1$ (Figure 2(d), expanded in Figure 4), was recorded at the Swiss Light Source, Paul Scherrer Institute Villigen, Switzerland, using an iPEPICO spectrometer [44]. The photon resolution was 0.004 eV and the step size 0.002 eV. A single progression is observed in $\nu_1$, the totally symmetric C–H stretching mode of $a_1$ symmetry, which peaks at $\nu=1$. This observation is independent of whether peak intensities or areas are measured. This spectrum has been recorded before by several groups [6-9], but never with an experimental resolution as good as 4 meV. Our peak positions and separations (Table 3) are in excellent agreement with those determined by Carlsson Göthe *et al.* from He II photoelectron spectroscopy [8], and we can unambiguously confirm their tentative observation that the linewidth increases as the vibrational quantum number in the $\nu_1$ mode increases. At a resolution of 4 meV, our signal-to-noise ratio is not good enough to determine peak positions or widths for $\nu \geq 4$, although data up to $\nu = 14$ are quoted by Carlsson Göthe *et al.* Since our experimental resolution is much narrower than the peak widths, deconvolution of the experimental resolution is not necessary, and we determine directly full width at half maximum (FWHM) values of 31, 52, 82 and 99 meV for $\nu = 0, 1, 2$ and 3, respectively. These values are roughly a factor of two smaller than those modelled by Carlsson Göthe *et al.*, and approximately correspond to the linewidths of the four vibrational levels of CH$_4^+$ $\tilde{A}$ $^2A_1$ ($\nu_1$). We therefore determine lifetimes of 21, 12, 8 and 6 fs, corresponding to dissociation rates of $4.8 \times 10^{13}$, $8.3 \times 10^{13}$, $1.2 \times 10^{14}$ and $1.7 \times 10^{14}$ s$^{-1}$, for $\nu = 0, 1, 2$ and 3, respectively. We note that the lifetimes are lower limits and the dissociation rates are upper limits, because it is assumed that all the peak broadening is due to the lifetime effect. In particular, we have ignored any effects due to instrumental resolution and rotational fine structure. *It seems very unlikely that the broadening is spectroscopic, and not dynamic, in nature, possibly due to unresolved vibrational polyads at higher energy caused by Coriolis and Fermi interactions, because these should be partially resolvable with a photon resolution of 0.004 eV.*

Threshold photoelectron photoion coincidence measurements have shown that the $\tilde{A}$ $^2A_1$ state of CH$_4^+$ dissociates directly, without prior internal conversion to the electronic ground state [7,9]. Internal conversion is slow due to the large Franck Condon gap of over 6 eV between the $\tilde{A}$ and $\tilde{X}$ states. Thus, the $\tilde{A}$ $^2A_1$ state behaves as an isolated state and dissociates non-statistically. The dominant fragment ion produced from the dissociation is
CH$_2^+$, the minority ions being CH$^+$ and CH$_3^+$, and the $\tilde{A}$ $^2$A$_1$ state of CH$_4^+$ correlates directly to CH$_2^+$ in C$_2v$ symmetry [47]; it also correlates to H$^+$ (+ CH$_3$) in C$_3v$ symmetry, but the H$^+$ signal is negligible in the experiment. Dutuit et al. have therefore suggested that the $\tilde{A}$ $^2$A$_1$ state of CH$_4^+$ dissociates directly to CH$_2^+$ + H$_2$ [7]. A sequential dissociation, (CH$_4^+$)* $\rightarrow$ (CH$_3^+$)* + H $\rightarrow$ CH$_2^+$ + H + H, cannot however be ruled out, and from approximate kinetic energy measurements on CH$_2^+$ is the preferred mechanism of Furuya et al. [9]. There is some additional evidence for this latter mechanism from the photoelectron spectrum of the CH$_3$ radical [48], where the first excited singlet electronic state of CH$_3^+$ is observed at 16.1 eV, i.e. 20.6 eV above the ground state of CH$_4$ [42]. Thus the first step of this two-step mechanism could be non-radiative dissociation of CH$_4^+$ $\tilde{A}$ $^2$A$_1$ into high vibrational levels of CH$_3^+$ $\tilde{A}$ $^1$E' which then decomposes to CH$_2^+$. Whichever mechanism is dominant, the data above give the first experimental measurements of upper limits for the dissociation rate producing CH$_2^+$ for the lowest four $\nu_1$ vibrational levels of CH$_4^+$ $\tilde{A}$ $^2$A$_1$.

5. Conclusions

Absolute cross sections and quantum yields for production of H$^-$, CH$^-$ and CH$_2^-$ from CH$_4$ over the energy range 12–35 eV have been determined. The signals of all three ions display a linear dependence with pressure, showing that they arise from an ion-pair mechanism and not from the multi-step process of dissociative electron attachment. The CH$^-$ and CH$_2^-$ spectra are observed here for the first time. Whilst the relative yield of H$^-$ is very similar to that observed in an earlier study by Mitsuke et al. [28,29], our cross section values are a factor of ca. 70 smaller than those quoted earlier. This discrepancy remains unresolved. The H$^-$ peaks are attributed to both direct and indirect ion-pair formation, whereas the CH$^-$ and CH$_2^-$ peaks probably arise from an indirect process, caused by ion-pair states crossing with doubly-excited Rydberg states. The threshold photoelectron spectrum of the second band of CH$_4$, ionisation to CH$_4^+$ $\tilde{A}$ $^2$A$_1$ at 22.40 eV, has been recorded with a resolution of 0.004 eV. The widths of the peaks observed for $\nu = 0$–3 in the $\nu_1$ vibrational ladder increase with $\nu$. They are the first direct measurement of a lower limit to the lifetime, and hence an upper limit to the unimolecular dissociation rate of these levels, into fragment ions.
Acknowledgments

We thank Dr David Shaw (Manager of beamline 3.1) for help with recording the anion spectra at the Daresbury synchrotron source. We thank Drs Andras Bodi and Melanie Johnson (SLS) and Ms Jonelle Harvey (University of Birmingham) for help with recording the TPES of CH$_4$ on beamline X04DB of the Swiss Light Source. The collaboration between the groups in Birmingham and Belfast was partially funded by EPSRC Network Grant No. GR/N26234/01. The STFC and EU are thanked for the provision of beamtime at the Daresbury synchrotron and the Swiss Light Source, respectively.

References


Table 1. AE$_{298}$ values, and absolute cross sections and quantum yields for production of anions from vacuum-UV photodissociation of CH$_4$.

<table>
<thead>
<tr>
<th>Anion</th>
<th>AE$_{298}$/eV$^a$</th>
<th>$\sigma_{\text{anion}}$(max)/cm$^2$$^b$</th>
<th>E/eV$^c$</th>
<th>$\Phi_{\text{anion}}$$^d$</th>
</tr>
</thead>
<tbody>
<tr>
<td>H$^-$</td>
<td>13.3 ± 0.1</td>
<td>1.4 x 10$^{-22}$</td>
<td>20.6</td>
<td>4.4 x 10$^{-6}$</td>
</tr>
<tr>
<td>CH$^-$</td>
<td>22.5 ± 0.2</td>
<td>5.9 x 10$^{-23}$</td>
<td>29.3</td>
<td>4.6 x 10$^{-6}$</td>
</tr>
<tr>
<td>CH$_2^-$</td>
<td>22.2 ± 0.2</td>
<td>2.8 x 10$^{-23}$</td>
<td>24.9</td>
<td>1.3 x 10$^{-6}$</td>
</tr>
</tbody>
</table>

$^a$ Appearance Energy (AE) at 298 K.

$^b$ Cross section for ion-pair formation at the peak maximum.

$^c$ Energy of peak maximum, at which $\sigma_{\text{anion}}$(max) and $\Phi_{\text{anion}}$ are determined.

$^d$ Quantum yields for anion formation, $\Phi_{\text{anion}}$, calculated from total vacuum-UV absorption cross sections of CH$_4$ taken from Ref. 24.
Table 2. Vibrational progressions observed in the yield of H\textsuperscript{−} from CH\textsubscript{4} (Figure 3).

<table>
<thead>
<tr>
<th>ν\textsubscript{1} vibrational state</th>
<th>Progression (a)</th>
<th>Progression (b)</th>
<th>Progression (c)</th>
</tr>
</thead>
<tbody>
<tr>
<td>0</td>
<td>E / eV</td>
<td>(ΔE / eV)</td>
<td>E / eV</td>
</tr>
<tr>
<td>0</td>
<td>20.00</td>
<td>(0.28)</td>
<td>21.24</td>
</tr>
<tr>
<td>1</td>
<td>0.20</td>
<td>21.52</td>
<td>22.32</td>
</tr>
<tr>
<td>2</td>
<td>20.52</td>
<td>(0.24)</td>
<td>21.76</td>
</tr>
<tr>
<td>3</td>
<td>20.76</td>
<td>22.00</td>
<td></td>
</tr>
<tr>
<td>4</td>
<td>22.24</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

\textsuperscript{a} The assignment of this peak to ν=0 is not definite, so the vibrational numbering of this progression is not certain.
**Table 3.** Full Width at Half Maximum (FWHM) values for vibrational peaks in the second photoelectron band of CH₄, ionisation to CH₄⁺ ̃A ²A₁ (Figure 4).

<table>
<thead>
<tr>
<th>ν₁ vibrational state</th>
<th>E / eV</th>
<th>(ΔE / eV)</th>
<th>FWHM / meV</th>
</tr>
</thead>
<tbody>
<tr>
<td>0</td>
<td>22.40</td>
<td>31 ± 3</td>
<td>(0.27)</td>
</tr>
<tr>
<td>1</td>
<td>22.67</td>
<td>52 ± 5</td>
<td>(0.26)</td>
</tr>
<tr>
<td>2</td>
<td>22.93</td>
<td>82 ± 8</td>
<td>(0.25)</td>
</tr>
<tr>
<td>3</td>
<td>23.18</td>
<td>99 ± 10</td>
<td></td>
</tr>
</tbody>
</table>
**Figure Captions**

**Figure 1:** Graph to determine the relative mass sensitivity of the Hiden Analytical HAL IV quadrupole mass spectrometer as a function of $(m/z)$. Sample gases included CF$_4$, SF$_6$, CF$_3$SF$_5$, CH$_3$F, CH$_3$Cl, CH$_3$Br, CH$_2$Cl$_2$, CF$_2$Cl$_2$, CFCI$_3$, C$_2$F$_4$ and c-C$_3$F$_8$. The mass spectrum of each sample was measured with 70 eV electron impact ionisation, and compared with the NIST spectrum [39]. At each $(m/z)$ value, the % yield from NIST is divided by the % yield from our spectrum, and the data are normalised to unity at $m/z$ 69 (i.e. CF$_3^+$). The squares show the data points, the solid line shows the best fit to a third order polynomial.

**Figure 2:** (a)-(c) Cross sections for H$^-$, CH$^-$ and CH$_2^-$ production following vacuum-UV photoexcitation of CH$_4$. Ion yields were measured between 12 and 35 eV with a step size of 0.1 eV and a wavelength resolution of 0.6 nm. Solid arrows show energies of thermochemical thresholds calculated for reactions (2)–(10), respectively. (d) Threshold photoelectron spectrum of CH$_4$ measured at a resolution of 0.004 eV using the imaging photoelectron photoion coincidence spectrometer at the Swiss Light Source [44]. The step size is 0.002 eV.

**Figure 3:** High resolution H$^-$ scan between 19.5 and 23.5 eV. The step size is 0.02 eV and the resolution is 0.25 nm, ca. 0.09 eV. The three progressions (a)-(c), have been assigned by Mitsuke et al. [28,29] as vibrational progressions in $v_1$ ($a_1$) within different Rydberg states converging on the $^2A_1$ state of CH$_4^+$. Progression (a) has been assigned as the $(2a_1)^1(3p)^1$ Rydberg state, (b) as the $(2a_1)^1(4p)^1$ Rydberg state, and (c) the $(2a_1)^1(5p)^1$ Rydberg state.

**Figure 4:** Expansion of Figure 2(d). Threshold photoelectron spectrum of the second band of CH$_4$, ionisation to CH$_4^+$ $^2A_1$ recorded with a step size of 0.002 eV and a resolution of 0.004 eV.
Figure 1
Figure 2

(a) H⁻

(b) CH⁻

(c) CH₂⁻

(d) TPES

Photon Energy / eV

Cross Section / 10⁻²³ cm²

TPE Signal / Arbitrary Units

(2), (3), (4), (5), (6), (7), (8), (9), (10)
Figure 3
Figure 4